Cloud microphysics Claudia Emde

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Introduction

Cloud microphysics

Contents

Reference

Atmospheric thermodynamics

Cloud formation and cloud dynamics

Cloud movies by B. Mayer

Cloud types

Hohe Wolken	Mittelhohe Wolken	Tiefe Wolken
11 Galerien - 225 Bilder	20 Galerien - 422 Bilder	10 Galerien - 207 Bilder
Vertikale Wolken	Optische Erscheinungen	Blitze
31 Galerien - 652 Bilder	8 Galerien - 153 Bilder	4 Galerien - 75 Bilder
-	-	
Regenbögen 6 Galerien - 103 Bilder	Sonstiges 15 Galerien - 299 Bilder	Niederschläge 12 Galerien - 239 Bilder
© 1998-2011: Bernhard Mühr, 10. April 2011		info@wolkenatlas.de
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http://www.wolkenatlas.de/

Rain drops



Figure 1] Topological changes of falling drops and fragmentation. Top row: series of events of the fragmentation of a $d_0 = 6$ mm water drop falling in an according testam of in at . The time interval between each image is $\Delta I = 4$, 7m. Stress factors the flattening of the drop into a pancake shape, the initiation of a bag bordered by a thicker corrugated irm, its break-up and the destabilization of the rim initiation of a single stress factors and the destabilization of the rim initiation disting to dispinet drops distributed in a size. Middle row as similar series defining the initial idments $\Delta O(B)$ are distributed and shape $\xi(r,t)$ and the distribution of the rim itself (highlighted in the instel), leaging to dispinet did drops size. Abstom row: the formation of a bags for narrandaror for the initial idment bag, which secures its bord D = 0 mm and $\Delta t = 7$ pms.

Villermaux et al. Nature Physics, 2009

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Ice crystals



In situ measurements of ice crystals in a tropical cirrus cloud

Heymsfield et al., Journal of Atmospheric Sciences, 2002

Cloud microphysics

Snow flakes



www.snowflakes.com by K.G. Libbrecht

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Energy balance of the Earth



IPCC report, 2007

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References

Atmospheric thermodynamics

Impact of clouds on climate change

Radiative Forcing Components



IPCC report 2007

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Introduction

Atmospheric thermodynamics

IPCC Bericht 2007



"Cloud feedbacks (particularly from low clouds) remain the largest source of uncertainty."

IPCC Report 2007, Technical Summary

Overview of cloud physics

- Atmospheric thermodynamics
- Microphysics of warm clouds
 - Nucleation of water vapor by condensation
 - Growth of cloud droplets in warm clouds (condensation, fall speed of droplets, collection, coalescence)
 - formation of rain
- Microphysics of cold clouds
 - homogeneous nucleation
 - heterogeneous nucleation
 - contact nucleation
 - crystal growth (from water phase, riming, aggregation)
 - formation of precipitation
- Observation of cloud microphysical properties
- Parameterization of clouds in climate and NWP models

- John M. Wallace and Peter V. Hobbs. *Atmospheric Science, An introductory survey*. Elsevier, 2006.
- R. R. Rogers. *A short course in cloud physics*. Pergamon Press, 1976.
- Hans R. Pruppacher and James D. Klett. *Microphysics of clouds and precipitation*. Springer, 1996.
- IPCC. Climate change 2007. Technical report, Intergovernmental Panel of Global Climate Change, 2007.

Additional publications and slides on website: www.meteo.physik.uni-muenchen.de/~emde/doku.php?id= teaching:cloud_microphysics:cloud_microphysics

Written exam in last week of semester: 9th February 2012, 14-16 h

Please let me know if date is inconvenient!

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Atmospheric thermodynamics

Gas laws and the ideal gas equation



Sir Robort Boyle (1627–1691)



Jacques Charles (1746-1823)

pV = mRT $p\alpha = RT$ $pV = nR^*T$ $p = n_0kT$



John Dalton (1766–1844) Images from Wikipedia





Ludwig Boltzmann (1844–1906)

Cloud microphysics

References

Atmospheric thermodynamics

The hydrostatic equation



$$\frac{\partial p}{\partial z} = -g\rho \qquad gdz = -\alpha dp$$



Sir Issac Newton (1642–1727) Image from Wikipedia

First law of thermodynamics



Fig. 3.4 Representation of the state of a working substance in a cylinder on a p-V diagram. The work done by the working substance in passing from P to Q is $p \, dV$, which is equal to the blue-shaded area. [Reprinted from *Atmospheric Science: An Introductory Survey*, 1st Edition, J. M. Wallace and P. V. Hobbs, p. 62, Copyright 1977, with permission from Elsevier.] Figure from Wallace and Hobbs

energy conservation

$$dq = du + dw$$

$$dq = du + pd\alpha$$

$$dq = c_p dT - \alpha dp$$

...

Adiabatic processes

adiabatic = change in physical state without heat exchange $\Rightarrow dq = 0$





$$dq = du + pd\alpha$$

T rises in adiabatic compression T=const. in isothermal process

 $T_C > T_B \Rightarrow p_C > p_B$